PATENT COOPERATION TREATY

PCT

INTERNATIONAL PRELIMINARY REPORT ON PATENTABILITY (Chapter II of the Patent Cooperation Treaty)

(PCT Article 36 and Rule 70)

pplicant's or agent's file reference 40936PC FOR FURTHER ACTION See Form PCT/IPEA/416							
International application No.	International filing date (day/month/year)	Priority date (day/month/year)					
PCT/AU2004/001080	12 August 2004	15 August 2003					
International Patent Classification (IPC)	or national classification a	and IPC					
Int. CI. 7 C25B 1/04, C01B 3/02, 3/	10						
Applicant PROTEGY LIMITED et al							
Examining Authority under Article 35	and transmitted to the ap						
2. This REPORT consists of a total of		over sheet.					
3. This report is also accompanied by A	· · · · · ·						
a. X (sent to the applicant and to	the International Bureau) a	total of 3 sheets, as follows:					
report and/or sheets con	sheets of the description, claims and/or drawings which have been amended and are the basis for this report and/or sheets containing rectifications authorized by this Authority (see Rule 70.16 and Section 607 of the Administrative Instructions).						
sheets which supersede earlier sheets, but which this Authority considers contain an amendment that goes beyond the disclosure in the international application as filed, as indicated in item 4 of Box No. I and the Supplemental Box.							
b. (sent to the International Bureau only) a total of (indicate type and number of electronic carrier(s)) containing a sequence listing and/or table related thereto, in computer readable form only, as indicated in the Supplemental Box Relating to Sequence Listing (see Section 802 of the Administrative Instructions).							
4. This report contains indications relat	ing to the following items:						
X Box No. I Basis of the repo	ort						
Box No. II Priority							
Box No. III Non-establishme	ent of opinion with regard t	o novelty, inventive step and industrial applicability					
Box No. IV Lack of unity of i	nvention						
X Box No. V Reasoned stater applicability; cita	ox No. V Reasoned statement under Article 35(2) with regard to novelty, inventive step or industrial applicability; citations and explanations supporting such statement						
Box No. VI Certain documents cited							
Box No. VII Certain defects i							
Box No. VIII Certain observat	Box No. VIII Certain observations on the international application						
Date of submission of the demand Date of completion of the report							
8 March 2005	· · · · · · · · · · · · · · · · · · ·	cember 2005					
Name and mailing address of the IPEA/AU		rized Officer					
AUSTRALIAN PATENT OFFICE PO BOX 200, WODEN ACT 2606, AUSTRA E-mail address: pct@lpaustralia.gov.au Facsimile No. (02) 6285 3929	ALIA DAV	ID K. BELL hone No. (02) 6283 2309					

Form PCT/IPEA/409 (Cover sheet) (January 2004)

International application No.

PCT/AU2004/001080

Во	x No. I Basis of the report					
1.	With regard to the language, this report is based on the international application in the language in which it was filed, unless otherwise indicated under this item.					
	This report is based on translations from the original language into the following language which is the language of a translation furnished for the purposes of:					
	international search (under Rules 12.3 and 23.1 (b))					
	publication of the international application (under Rule 12.4)					
	international preliminary examination (under Rules 55.2 and/or 55.3)					
2.	With regard to the elements of the international application, this report is based on (replacement sheets which have been furnished to the receiving Office in response to an invitation under Article 14 are referred to in this report as "originally filed" and are not annexed to this report): the international application as originally filed/furnished					
	X the description:					
	pages 1 to 10 and 12 to 16 as originally filed/furnished					
	pages* 11 received by this Authority on 8 March 2005 with the letter of 8 March 2005					
	pages* received by this Authority on with the letter of X the claims:					
	pages as originally filed/furnished					
	pages* as amended (together with any statement) under Article 19					
	pages* 17 and 18 received by this Authority on 15 November 2005 with the letter of 15 November 2005					
	pages* received by this Authority on with the letter of the drawings:					
ı	pages 1 and 2 as originally filed/furnished pages* received by this Authority on with the letter of pages* received by this Authority on with the letter of					
	a sequence listing and/or any related table(s) - see Supplemental Box Relating to Sequence Listing.					
3.	The amendments have resulted in the cancellation of:					
	the description, pages					
	the claims, Nos.					
	the drawings, sheets/figs					
•	the sequence listing (specify):					
	any table(s) related to the sequence listing (specify):					
4.	This report has been established as if (some of) the amendments annexed to this report and listed below had not been made, since they have been considered to go beyond the disclosure as filed, as indicated in the Supplemental Box (Rule 70.2(c)).					
	the description, pages					
	the claims, Nos.					
	the drawings, sheets/figs					
	the sequence listing (specify):					
	any table(s) related to the sequence listing (specify):					
•	If item 4 applies, some or all of those sheets may be marked "superseded."					

International application No. PCT/AU2004/001080

ID 200 4040707 (day/month/year) (day/month/year)	. Certain published documen	Certain published documents (Rule 70.10)				
JP 2004210597 29 July 2004 Unavailable 6 January 2003 2. Non-written disclosures (Rule 70.9) Kind of non-written disclosure Date of non-written disclosure (day/month/year) Date of written disclosure referring to non-written disclosure	Patent No.		Filing date (<u>day/month/year)</u>	Priority date (valid clain		
Kind of non-written disclosure Date of non-written disclosure Date of written disclosure (day/month/year) Date of written disclosure	JP 2004210597	29 July 2004		6 January 2003		
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Supplemental Box

In case the space in any of the preceding boxes is not sufficient.

Continuation of: V

In addition, Claim 1 is merely a reiteration of the laws of thermodynamics and kinetics. <u>Irrespective</u> of how the water is finally dissociated the following is an expression of how the reaction proceeds with an increase in temperature. It clearly shows that an increase in temperature will increase the number of dissociated H₂O molecules at or near the reactive or catalytic surface. In particular the reaction:

$$H_2O \rightarrow H_2 + \frac{1}{2}O_2$$

has an enthalpy of + 285.83 KJ/mol. Hence it is endothermic. Consequently the reaction follows the well known law of endothermic reactions, that is: "increased temperature favours the products". This is outlined in D13. The additional "restriction" introduced by way of Article 34 amendments, that "energy is added by the addition of steam instead of the energy provided as an applied electrical current" is merely substituting one form of energy for another and as a consequence the aforementioned and well known laws of "hermodynamics still apply and there is no change in the scope of the claims. In the simplest terms the amount of energy required to convert:

$$H_2O \rightarrow H_2 + \frac{1}{2}O_2$$

remains constant irrespective of the pathway. Documents D5, D13 and D15 explicitly recognise this fundamental law of thermodynamics. The other documents take advantage of and implicitly recognise this law.

It is well understood that the reaction rate increases with an increase in temperature, again as outlined in D13. Documents D1 to D4 and D6 to D12 clearly disclose the application of these principles in the electrolysis of water, by elevating the temperature (in most cases to well above the boiling point of water) of the reactants during electrolysis. Consequently it is considered that claim 1 does not contain an element of novelty, nor does it involve an inventive step. It should also be noted that by definition a catalyst cannot change the equilibrium of the reaction. A catalyst simply increases the rate at which the reaction proceeds. Simply using steam to heat water irrespective of the catalyst (or reactive surface) used, or the final reaction temperature obtained is unlikely to produced a significant increase in the quantities of hydrogen fuel produced.

Document D14 by the applicant introduces the concept of combining two half cell reactions, with the idea of increasing hydrogen production. The teachings from this document may be combined with any of $_{\sim}$ occurrents D1- D13 and D15 to produce a method of producing $_{\sim}$ of the sort disclosed by D14 with an "increased number of dissociated $_{\sim}$ 0 molecules" near the reactive surface. Consequently claims 2 to 11 do not involve an inventive step.

The invention as defined in the present claims is therefore not Novel and does not have an Inventive Step when compared to the disclosures of the cited documents D1 to D15. The invention is industrially applicable.

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Box No. V Reasoned statement under Article 35(2) with regard to novelty, inventive step or industrial applicability; citations and explanations supporting such statement

						
1.	Statem	nent				
		Novelty	(N)	Claims	YES	
i ,				Claims 1 to 11	NO	
		Inventiv	e step (IS)	Claims	YES	
				Claims 1 to 11	. NO	
		Industria	al applicability (IA)	Claims 1 to 11	YES	
				Claims	NO	
2.	Citation	ns and ex	xplanations (Rule 7	0.7)	·	
	D1	= .	WO 2000/017	418 A1 (The Regents of the Univers	elty of California) 20 March 2000	
	D2	=	GB 1490650 A	(Blum P.), 2 November 1977	sty of California), 30 March 2000	
	D3	=	GB 2010333 A	(Kernforschungsanlage Julich Gm	hH) 27 June 1070	
	D4	=	WO 1994/012	690 (Lasich J), 9 June 1994	511), 21 Julie 1919	
	D5	=	EP 1006078 A	1 (Yosohiro S et al.), 7 June 2000		
	D6	=	DE 3101210 A	(1 (Dornier System GmbH.), 29 July	1982 & Derwent Abstract Accession	
	No. 63841 E/31, Class E36, J03 D7 = DT 2549471 A1 (Dornier System GmbH), 12 May 1977 & Derwent Abstract Ac					
110. 34043				J Class E36, J03		
		_	(Pamiraz Carr	act Accession No. 2000- 497491/44	, Class E36, J03, PT 102238 A	
	Ð9	(Ramirez Garcia J.), 31 July 2000 Derwent Abstract Accession No. 47774B/26, Class E36, J03, JP 54-061088 A (Hitachi				
			Shiphid Eng K	K), 17 May 1979	s E36, J03, JP 54-061088 A (Hitachi	
	D10	=	Denvent Aber	of Accession No. 427758/98 Clas	- F00 100 ID 54 004005 - 4444 - 1	
	. 7:-	*	Shipbid Fna K	K), 17 May 1979	s E36, J03, JP 54-061089 A (Hitachi	
	D11	=		act Accession No. 33534 K/14, Clas	20 E26 IO2 ID E0024400 A	
		=	(Arakawa T), 2	8 February 1983	58 E30, JUS, JP 58034183 A	
	D12	=	Derwent Abstr	act Accession No. 94-084061/11 C	lass X25, CN 1072465 A (Zheng J),	
			ZU Way 1993			
	D13	=	"Physical Cher	nistry" 3rd Ed. P.W. Atkins, Oxford L	Iniversity Press 1986	
			15BN 0-19-855	196-7.	•	
	D14	=	WO 2000/0706	99 (Protegy Limited), 23 November	r 2000	
	D15	=	-HYDROGEN	PRODUCTION BY WATER SPLITT	ING USING MIXED CONDUCTING	
			MEMBRANES	U. Balachandran, T. H. Lee, S. Wa	ing, and S. E. Dorris February 2003,	
			Manuscript to I	e submitted for publication in the D	maria all control contains 2000,	

The applicant's attention is directed to the document D5. This document clearly and unmistakeably discloses a process for generating hydrogen gas through direct thermal decomposition of water in which steam is used to heat water which is then passed under positive pressure and elevated temperature over a catalyst (zeolite) that separates hydrogen from the heated water. The invention defined in claims 1 to 11 is therefore not novel and does not have an inventive step when compared to the disclosure of D5.

Manuscript to be submitted for publication in the Proceedings of the National Hydrogen Association's 14th Annual U.S. Hydrogen Meeting, Washington, DC, March 4-6, 2003.

Continued on supplemental sheet